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CHAPTER 12 REVIEW Solutions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Solid CaCl 2 does not conduct electricity. Explain why it is considered to be an electrolyte. 2. Explain the following statements at the molecular level: a. Generally, a polar liquid and a nonpolar liquid are immiscible. b.

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CHAPTER . 12 . REVIEW '''-- Solutions . SHORT ANSWER Answer the following questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. 3. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCI and adds I L of water to the solid. James made several errors.

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CHAPTER 12 REVIEW Solutions SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. The following are statements about the dissolving process. Explain each one at the molecular level. a. Increasing the pressure of a solute gas above a liquid solution increases the CHAPTER 12 REVIEW Solutions - Manasquan Public Schools Start studying

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